## Effect of Pt Buffer on Magnetic Property of Fe<sub>4</sub>N Thin Film Deposited by Atmospheric Pressure Halide CVD Method

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Polycrystalline Fe4N thin films were prepared by AP-HCVD from the starting materials of FeCl3 and NH3. The effect of Pt buffer layer on Si(100) substrate was examined. It was clarified that coexistence of the second phases such as Fe3N and  $\alpha$ -Fe was detected on Pt buffer layer. The sample was deposited on 10 nm-thick Pt buffered Si(100) substrate. It was considered that the introduction of Pt buffer layer affected to the coverage and uniformity of the film.

Key words: Fe<sub>4</sub>N, AP-HCVD, Pt-buffer layer, thin film, Si substrate

## 1. INTRODUCTION

Table 1 Course

Various iron nitrides such as  $\gamma$ '-Fe<sub>4</sub>N (cubic),  $\zeta$ -Fe<sub>2</sub>N (orthorhombic),  $\epsilon$ -Fe<sub>2</sub>N (hexagonal), ε-Fe<sub>3</sub>N (hexagonal) and  $\alpha$ "-Fe<sub>16</sub>N<sub>2</sub> (body centered cubic) are known [1]. All of these iron nitrides are interstitial compounds. It is also known that several iron nitrides such as  $\gamma$ '-Fe<sub>4</sub>N and  $\alpha$ ''-Fe<sub>16</sub>N<sub>2</sub> have high saturation magnetization [2]. Crystal system and saturation magnetization for several iron nitrides were shown as Table 1. Among these iron nitride, Fe<sub>4</sub>N is promising due to its mechanical strength and high oxidation resistance [3]. Crystal structure of Fe<sub>4</sub>N is shown in Figure 1. In this work, preparation of Fe<sub>4</sub>N thin films on Si(100) substrate by atmospheric pressure halide CVD (AP-CVD) was examined. In this work, the effect of Pt buffer later on the crystal growth and magnetic property of Fe<sub>4</sub>N was examined.

## 2. EXPERIMENTAL

Figure 2 shows schematic drawing of the AP-CVD equipment used in this work. The equipment is made of silica glass and composed of a reactor, a substrate rod, a substrate folder and a boat in which raw material (FeCl<sub>3</sub> is charged). In the reactor, carrier gas (N2), counter gas  $(N_2)$  and blower gas  $(NH_3+N_2)$  were provided. The size of the reactor is 1,010 mm long and \$22 mm in internal diameter. The reactor is covered with two electric furnaces; one is used to evaporate raw material (FeCl<sub>3</sub>) and another is used to deposit iron nitride films. Growth conditions of Fe<sub>4</sub>N were summarized in Table 2. In this work, the films were deposited on Si(100) substrate with natural oxide and Si(100) substrates covered with 10 and 70 nm-thick Pt buffer layers. The films were characterized by x-ray diffraction, scanning electron microscopy (SEM) and vibration sample magnetometer (VSM).



Figure 1. Crystal structure of Fe<sub>4</sub>N.

|   |                           |                                  | _ |  |  |
|---|---------------------------|----------------------------------|---|--|--|
| al system and saturation magnetization for several iron nitrides. |                           |                                  |   |  |  |
| Composition   | Crystal structure         | Saturation magnetization (emu/g) |   |  |  |
| $\zeta$ -Fe <sub>2</sub> N  | Orthorhombic              | -                                |   |  |  |
| $\epsilon$ -Fe <sub>3</sub> N                                     | Hexagonal                 | 138                              |   |  |  |
| $\gamma'$ -Fe <sub>4</sub> N                                      | Cubic                     | 182                              |   |  |  |
| $\alpha$ "-Fe <sub>16</sub> N <sub>2</sub>                        | Breast conserving therapy | 240                              |   |  |  |

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Figure 2. Schematic drawing of the AP-CVD equipment.



Figure 3. XRD patterns of Fe4N thin films on Pt buffered Si substrate.

a) Si(100) without Pt, b )Si(100) with 10 nm-thick Pt, c) Si(100) with 70 nm-thick Pt.

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|------------------------------------|--|--|
| Source                             | FeCl <sub>3</sub> ,NH <sub>3</sub>   |  |
| Substrate                          | Pt-coated Si(100)  |  |
| Substrate place                    | Under NH <sub>3</sub> blow   |  |
| Pt thickness [nm]                  | 0, 10, 70  |  |
| FeCl <sub>3</sub> partial pressure |  |  |
| [atm]                              | 7.63×10 <sup>-6</sup>  |  |
| NH <sub>3</sub> partial pressure   | 1  |  |
| [atm]                              | $1.42 \times 10^{-1}$  |  |
| Growth temperature                 |  |  |
| [°C]                               | 600  |  |
| V/VIII                             | 1.87 ×104  |  |
| Carrier gus                        | N2   |  |
| Growth time [min]                  | 30   |  |
| Total flow late [sccm]             | 900  |  |

Table.2. Growth conditions of Fe<sub>4</sub>N.

## 3. RESULTS AND DISCUSSION

Figure 3 shows XRD patterns of the deposited films on (a) Si(100) substrate without Pt buffer layer, (b) and (c) Si(100) substrate with 10 and 70 nm-thick Pt buffer layers, respectively. This figure indicates that polycrystalline phase-pure Fe<sub>4</sub>N thin film was deposited on Si(0100) without Pt buffer layer. On the other hand, trace Fe<sub>3</sub>N was detected for the film deposited on 10 and 70 nm-thick Pt buffer layer. On 70 nm-thick Pt,  $\alpha$ -Fe was also detected. The  $\alpha$ -Fe would be caused by the reduction of Fe<sub>4</sub>N by H<sub>2</sub> that is derived by the decomposition of NH<sub>3</sub> due to the catalytic action of Pt.

Figures 4 and 5 show SEM photographs of surface and cross section of the films, respectively. These figures indicate that the films were granular and the surface is significantly rough. It is also observed that the grain size of Fe<sub>4</sub>N film deposited on Pt buffer layer is smaller than that deposited on Si(100) without Pt buffer layer. This tendency agreed well with the crystallite diameter measured by the Schrrer's equation as shown in Table 3.



Fig 4. Surface SEM images of Fe<sub>4</sub>N thin films on Pt buffered Si substrate a) Si(100) without Pt, b)Si(100) with 10 nm-thick Pt.

Figure 5. Cross-section SEM images of Fe4N thin films on Pt buffered Si substrate a) Si(100) without Pt, b)Si(100) with 10 nm-thick Pt.

| Table.3. Crystalline size of | of $Fe_4N$ thin j  | film on the basis of Scherrer's equation. |
|------------------------------|--|---|
|                              | and a second sec |   |

| Sample                      | Particle diameter (nm) |  |
|-----------------------------|------------------------|--|
| Si(100) without Pt          | 146                    |  |
| Si(100) with 10 nm-thick Pt | 106                    |  |

Table.4. Saturation magnetization and coercive force of Fe<sub>4</sub>N thin film.

| Sample                      | Saturation magnetization (emu) | Coercive force (Oe) |
|-----------------------------|--------------------------------|---------------------|
| Si(100) without Pt          | 0.238                          | 81                  |
| Si(100) with 10 nm-thick Pt | 0.390                          | 52                  |

Figures 6 and 7 show M-H curves of  $Fe_4N$  films deposited on Si(100) without and with Pt buffer layers, respectively. Saturation magnetization and coercive force were shown in Table 4. Since it was difficult to estimate the thickness of the film, saturation magnetization is not normalized as emu/g. However, from the shape of M-H curves, it can be mentioned that coercive force of  $Fe_4N$  deposited on Pt is smaller than that deposited on Si. By comparison with figure 3, it would be considered that the difference of magnetic property, especially coercive force, is derived by the difference of crystallinity.



Fig. 6 M-H curves of Fe4N thin film deposited on Si(100) substrate without Pt buffer layer.



Fig. 7. M-H curves of Fe4N thin film deposited on Si(100) substrate with 10 nm-thick Pt buffer layer.

4. REFERENCES

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