

Experiments of Creating Carbon Nanotubes in Liquid Helium

T. Shigematsu, T. Nishimoto, H. Kawasaki, Y. Johno, T. Ohshima, S. Suetake,
K. Nakashima, Y. Yagyu and Y. Suda

SASEBO National College of Technology, 1-1 Okishin, Sasebo, Nagasaki 867-1193, JAPAN

Fax: 81-956-34-8490, e-mail: shige@post.cc.sasebo.ac.jp

Arc in liquid method has been developed as a cost-effective technique to fabricate various kinds of carbon nanomaterials. In liquid nitrogen, especially high-quality multi-wall carbon nanotubes were observed. So, our research aims at creating carbon nanomaterials using contact arc method in liquid nitrogen and in liquid helium. For this research, a special evaporation cryostat, which has moving parts at low temperature part, is prepared. Experiments in liquid nitrogen were carried out at current density $8\text{kA}/\text{cm}^2$, $10\text{kA}/\text{cm}^2$, $12\text{kA}/\text{cm}^2$ and $14\text{kA}/\text{cm}^2$. At current density 10 and $12\text{kA}/\text{cm}^2$ ample fibrous carbon nanomaterials could be obtained. Then, based on the results of discharged experiments in liquid nitrogen, productive experiments were done in liquid helium. At that time, in discharged experiments, ample carbon clusters could be observed at current density $10\text{kA}/\text{cm}^2$. Using TEM Imaging, it turns out that these clusters were multi-wall carbon nano-tubes. This is the first observation of creating carbon nanotubes in liquid helium.

Key words: Low temperature liquid, CNT, Contact arc method

1. INTRODUCTION

In the past few years, researches on carbon nano-tubes have been developed remarkably [1]-[4]. Among those, researches on the characteristics of carbon nanotube (CNT) have been done, so that the electronic conditions and electrical characteristics have been cleared[5]. As effective production method to synthesize nanotube stably and abundantly, contact arc method, laser ablation method, and chemical vapor deposition method have been proposed. The disadvantage of these methods, however, is to require expensive machinery. Recently, as simplified carbon arc nanotube synthesis method, it has been demonstrated that carbon nanomaterials can be synthesized by arc discharge generated in liquid water[6] or in liquid nitrogen[7]. Not only this method is easy to operate but also it allows to produce high-quality multi-walled carbon nanotubes at high production rates[7] and to produce some new type of nanomaterials (nanooxion)[6]. Here, Shigematsu et al. shows that in low temperature liquid carbon atoms emitted by discharge are cooled down quickly and start to combine with some amount of energy, it is possible to create carbon nano materials with basic structures[8][9]. Then, in order to challenge to find some new carbon nanomaterials and to create high-quality carbon nanomaterials, the aim of this research is to investigate the effectiveness and possibility of the production of carbon nanomaterials using arc discharge in low temperature liquid, such as liquid helium. As the pre-stage experiment, so as to find out the creation condition, discharged experiments were

carried out in liquid nitrogen.

2. EXPERIMENTAL SETUP

In our present experiment, contact arc method is adopted, and original instruments are designed and produced. Fig.1 shows the outline of our experimental system. The target carbon rod is set in low temperature liquid and it is high charged by an outside electric source. Then, the charged carbon rod can be controlled its discharge period by electromagnetic coil at room temperature. The process of creation is observed through investigating combine spectrum of carbon materials. An ICCD camera (HAMAMATSU C5909) for analyzing spectrum is set at room temperature. Fig.2 shows our experimental instruments. This instrument is vacuum-jacketed structure, and has an evaporation cryostat, which consists of liquid nitrogen dower and liquid helium dower in order to prevent evaporation of liquid. So as to remove

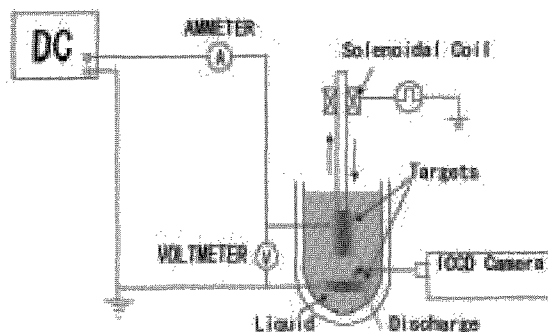


Fig.1 Schematic diagram of contact arc method

experimental cells easily, quick coupling method is utilized for the instrument's top plate. Because most of arc energy is absorbed by the liquid and it causes extraordinary evaporation, the instrument has a leak valve of 10 lit./sec as a solution.

The dielectric breakdown voltage in liquid helium is approximately 20kV at 1mm[10], and it discharges 20 more percent[11]. That's why, for contact arc method, it is necessary to let carbon rod electrodes approach carefully. Considering this, a system which allows electrode to move slowly is prepared using bellows and isolators (see Fig.4).

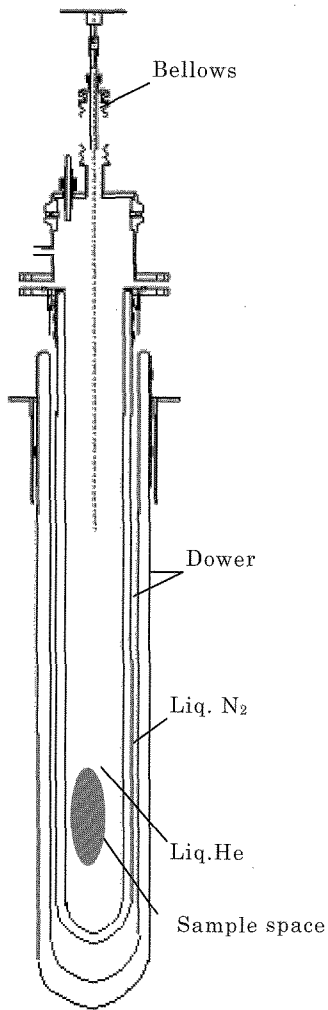


Fig.2 Our original experimental instrument.

An experimental cell used in contact arc method is shown in Fig.3. The upper carbon rod is made to move up and down. Two carbon rod electrodes, which are purchased from Nilaco Coro., Japan: 99.99% purity, 10mm diameter and 30mm long) are perpendicularly placed in liquid helium.

As for the confirmation of product after arc experiment, SEM and TEM observations are carried out analyzing the collector located at the lower part of experimental cell.

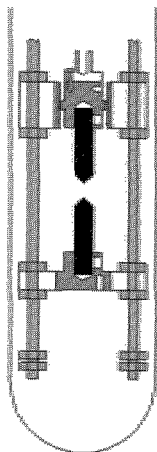


Fig.3 Experimental cell for contact arc method in liquid experimental instrument.

3. EXPERIMENTAL RESULTS

3-1. Experiment in liquid nitrogen

The experiments were carried out at current density 8kA/cm², 10kA/cm², 12kA/cm² and 14kA/cm². The time of arc was 1 second. Every time arc was done, the resultant product was collected. Fig.4 shows the result of spectroscopic analysis of discharged experiments in liquid nitrogen. In Fig.4, when current density is 8kA/cm², every experiment shows almost the same result. As for spectroscopic analysis, all lights in discharged experiments are integrated. Because C₂ swan band can be observed, it turns out that carbon ions emitted by discharge combines and form carbon nano clusters. In Fig.5 to Fig.8, each of SEM pictures is shown. When current density was less than 8kA/cm², ample carbon nanomaterials were not obtained. On the other hand, when current density was 10kA/cm² and 12kA/cm², many fibrous carbons were obtained. Then, when current density was 14kA/cm², the production rate seemed to decrease.

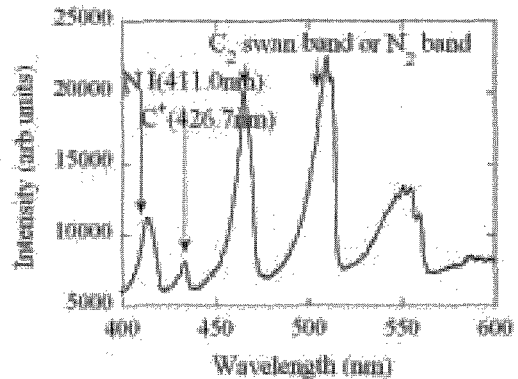


Fig.4 The result of spectroscopic analysis of discharge experiments in liquid nitrogen. Evaporations of carbon and C₂ swan band of clusters are observed.

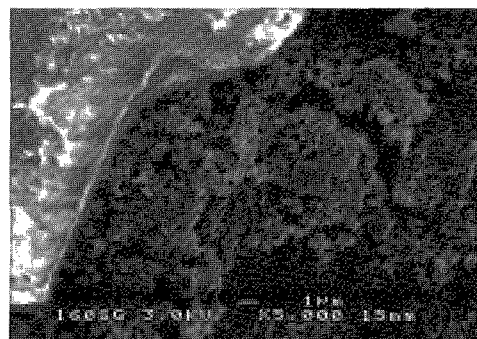


Fig.5 SEM image of arc in liquid nitrogen, current density 8kA/cm². Carbon nanomaterials were not obtained.

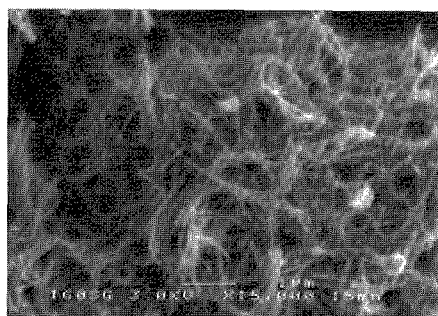


Fig.6 SEM image of arc in liquid nitrogen, current density $10\text{kA}/\text{cm}^2$. Many fibrous carbons were obtained.

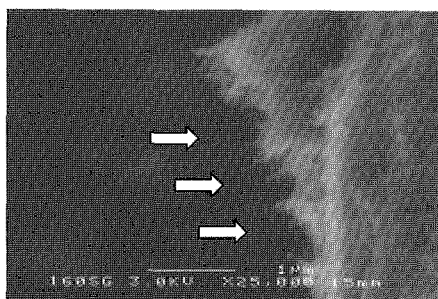


Fig.7 SEM image of arc in liquid nitrogen, current density $12\text{kA}/\text{cm}^2$. Many fibrous carbons were obtained. Arrow heads in Figures point out fibers

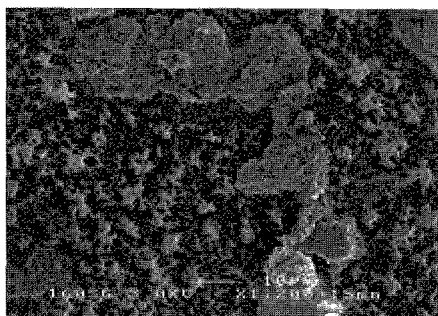


Fig.8 SEM image of arc in liquid nitrogen, current density $14\text{kA}/\text{cm}^2$. Carbon nanomaterials were not obtained.

At low electric density, the small amount of carbon evaporations can be one of the reasons why clusters were not observed. On the other hand, at high electric density ($14\text{kA}/\text{cm}^2$), most of energy caused by discharge is used for the brittle fracture at low temperature region and it prevents carbon from evaporating. Arrow heads in Figure 7 point out fibers, which can be carbon materials.

(In Figures, the foundation, on which carbon fibers are planted, is thought to be some mixture of broken graphite pieces caused by arc)

3-2. Experiment in liquid helium

Fig.9 to Fig.12 show SEM and TAM images

of contact arc experiments in liquid helium and spectroscopic analysis under the condition that a current was $10\text{kA}/\text{cm}^2$.

Because the spectrums of carbon ion, HeI, and HeII were observed at the time of this experiment, it is clear that emitted carbon atoms lost their energy due to Ionization loss process. It is also clear that carbons were undoubtedly emitted into liquid by discharge. However, C_2 swan band was not observed. This shows that carbon clusters were little and spectrums were not strong enough for the ICCD camera to observe them. Fig.10 and Fig.11 show the result of SEM observation of discharge experiment in liquid helium. In them, needle-shape carbon clusters can be observed. Their diameters range approximately from 10nm to 30nm and some of them are longer than $1\ \mu\text{m}$. Then, Fig.12 shows the TEM images of Fig.11. The images show that needle-shape carbon clusters are tube and have multi-walls. Though the difference of diameter is related to the number of walls, it is not possible to find out how many walls these clusters have because of the low sensitivity of TEM imaging.

Fig.13 shows one of these clusters. Though its inside diameter is $2\sim 4\text{nm}$, its outside diameter is approximately 20nm . This shows that it has multi-walls.

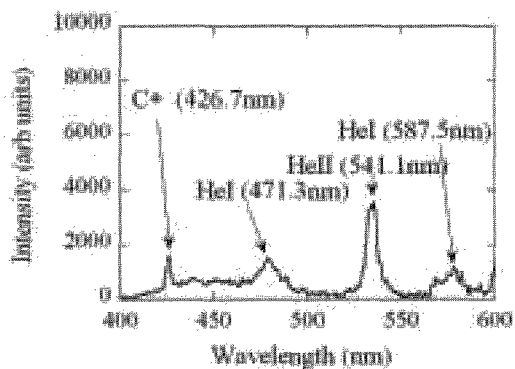


Fig.9 Representative result of spectroscopic analysis in liquid helium

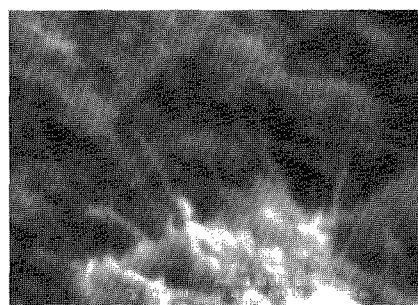


Fig.10 SEM images of the nano-material products using contact arc experiment in liquid helium. Ample needle-shape carbon nanomaterials were observed.

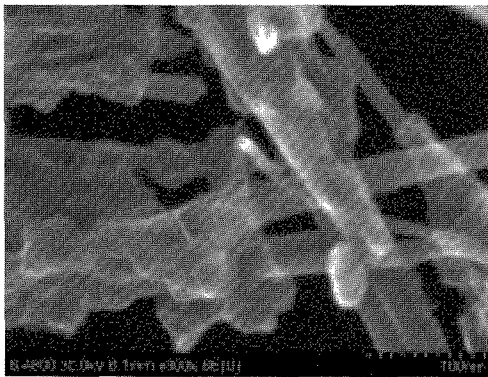


Fig.11 FESEM images of the nano-material products using contact arc experiment in liquid helium.



Fig.12 TEM images of the nano-material products using contact arc experiment in liquid helium. This figure shows the TEM images of Fig.11. The images show that needle-shape carbon clusters are tube and have multi-walls.

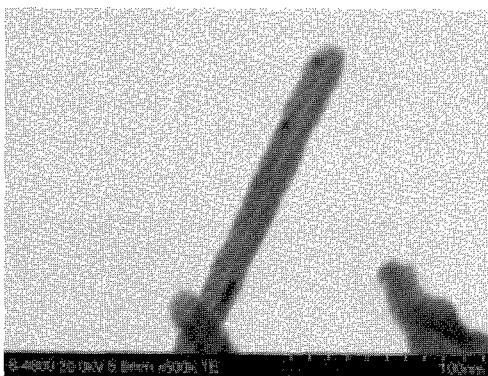


Fig.13 TEM images of the nano-material products using contact arc experiment in liquid helium. Though its inside diameter is 2~4nm, its outside diameter is approximately 20nm. Unfortunately, it is not possible to find out how many walls these clusters have because of the low sensitivity of TEM imaging. This shows that it has multi-walls.

4. CONCLUSIONS

In this research, discharge experiments have been carried out in low temperature liquid. In liquid nitrogen, fibrous carbon nanomaterials were obtained at current density $10\text{kA}/\text{cm}^2$ and $12\text{kA}/\text{cm}^2$. As for arc in liquid helium method at current density $10\text{kA}/\text{cm}^2$, carbon nanomaterials could be obtained for the first time. The TEM imaging shows that they were multi-wall carbon nanotubes. It also shows that the top has a ball-shape as if fullerene were attached.

6. ACKNOWLEDGMENTS

The authors wish to thank Prof. A. Grishin of Kungliga Tekniska Högskolan in Sweden for his helpful suggestion

7. REFERENCES

- [1] S.Ijima, Nature, 354, 56, (1991)
- [2] A.G.Rinzler et al., Science, 269, 1550, (1995)
- [3] W.A.Heer et al., Science, 270, 1179, (1995)
- [4] P.G.Collins, A.Zettl, Phys. Rev. B, 55, 9391, (1997)
- [5] M.S.Dresselhaus et al., "Science of Fullerenes and Carbon Nanotubes", Academic Press, New York, NY, San Diego, CA (1996)
- [6] N.Sano, H.Wang et al., Nature, 414, 506, (2001)
- [7] M.Ishigami, J.Cummings et al., Chem, Phys. Lett., 319, 457, (2000)
- [8] T.Shigematsu et al., Trans. Mater. Res. Soc. Jpn., 32[1], 167 (2007)
- [9] T.Shigematsu et al., Abstracts of CSJ Conference, 75, 148 (2006)
- [10] M.Hara et al., Cryogenics, 29(4), 448 (1989)
- [11] M.Hara et al., IEEE Trans. Elec. Insul., RI-23(4), 769 (1988)

(Received :January 15, 2008; Accepted June 20, 2008)